

Bridge Preservation

Corrosion Mitigation Option in Concrete Repair Systems

September 29, 2010 Northeast Bridge Preservation Partnership Meeting TSP-2

TODAY WE WILL DISCUSS

- Corrosion of Bridge Decks
- Surface Applied Corrosion Inhibitors
- Advances in Surface Applied Corrosion Inhibitors
- Extension of Service Life of Bridge Decks using Advanced Corrosion Mitigation Technology



CORROSION PROCESS



Electrochemical Corrosion Reactions



Important Takeaways:

- Chlorides are not consumed in the process
- Reinforcing steel acts as both the "Anode" and "Cathode"
- [H+] produced drops pH and further deteriorates the passivation layer
- The electrolyte is the ion-rich pore solution in the concrete
- Anode-Cathode reactions occur at the same rate
- Need both oxygen and water to complete the reaction

WHAT IS A SURFACE APPLIED CORROSION INHIBITOR?

- **Clear Penetrating Liquid**
 - Low Viscosity
 - Low Surface Tension
 - Reactive (Concrete / steel)
- Application
 - Bridge Decks
 - Piles
 - Girders
 - Jersey Barriers
- Spray or Roller Applied



VARIOUS TECHNOLOGIES



Conventional protection methods may prevent corrosion, but they are not as effective on active corrosion



Silanes: Made with Different Functionality



FHWA RD 98-153 Crack Beam Corrosion Study

- Based on FHWA Sponsored Research
- 0.47 w/c Concrete Slabs 12" x 12" x 7"
- 5/8 inch Diameter Black Bars
- One Inch Clear Cover
- Two 12 mil Cracks, One Inch Deep
- Cracks Run Along Top Bars



TEST CONDITIONS

- 48 Weeks Cyclic Salt Water Ponding
- 15% NaCl Solution
- Four Days Ponding with Salt Water @ RT
- Three Days Drying at 100°F
- Humidity Maintained at 60% to 80%



Scenario 1: PRESERVATION "New Concrete"



RESULTS PRESERVATION "New Concrete"



Scenario 2: RESTORATION Existing Corrosion in Concrete



RESULTS RESTORATION Existing Corrosion in Concrete



FHWA RD 98-153 "Autopsy"



FHWA SUMMARY Performance on Cracked Concrete

99% Effective in Preventing Corrosion in New Construction

92% Effective in Reducing Active Corrosion

93% More Effective Than Epoxy Coated Steel



FHWA Protocol Summary

	Reduction	Increase in	Reduction in ¹ / ₂
	In Corrosion	Resistance	Cell Potential
New Construction	99%	1090%	71%
Existing Corrosion	92%	386%	40%

COMPARATIVE DATA Cracked Specimens

Corrosion Rate on Cracked Specimens for Different Systems Using FHWA-RD-98-153 Protocol

- Black Bars -- 4053 mV
- Epoxy, 0.004% damage -- 325 mV
- Epoxy, 0.5% damage -- 971 mV
- Copper-clad -- 111 mV
- 316 SS -- 5 mV
- ORGANOSILANE CORROSION INHIBITORS -- 58 mV

FIELD EVALUATIONS Imbedded Reference Anodes **Continuous Monitoring**



Measurement provides ideal corrosion protection for steel, since the alkalinity of the concrete in combination with water and oxygen of a Corrosion of a uniprotective oxide film on the steel surface (Figure 1(a)). This passive film decreases the corrosion rate of steel to virtually zero and increases the durability of steel reinforced Inhibitor Through Online Monitoring years. It is well known that such natural corrosion protection of steel can be compromised by chloride ions and other sub-1[b]). In this case, significant corrosion may occur, causing decreased load capac-

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Typical repair of reinforced concrete structures showing corrosion damage involves removal of carbonated or chloride-contaminated concrete and subsequent retrofilling with new concrete. Corrosion inhibitors that can be applied by spraying onto the

Chlorides, often originating from deicing salts, have caused significant corrosion damage on reinforced concrete structures. Also, carbonation-the reaction of carbon dioxide (CO,) with concrete-may decrease the alkalinity of the concrete and activate corrosion.

The combination of steel and concrete

causes the formation of a thin protective

concrete structures to more than 100

stances that penetrate into the concrete and diffuse to the steel surface (Figure

ity in the structure. Under certain conditions, corrosion rates of up to 0.7 mm/y

CASE HISTORY Protecting Steel in Concrete

may occur.1

Typically, repair for such corrosion involves the removal of the carbonated or chloride-contaminated concrete, followed



PERFORMANCE Field Evaluation Silane Based Inhibitor



REPASSIVATION OF STEEL Reaction with passivation layer

This is the nitrogen area of the curve. The admixture contains no amines, while the surface applied Corrosion Inhibitor has aminosilanes. This accounts for the higher intensity of the surface applied inhibitor.



FIELD PERFORMANCE

TEST IT!!!

It is always better to protect than to repair!





- July 2009 untreated testing performed by the University of Waterloo, Toronto
 - Corrosion rates were low although evidence of corrosion was apparent
 - A section was then treated with silane based SACI
- September 2009 measurements were taken on both treated and untreated areas
- Where SACI was applied, the resistivity of the concrete deck increased by 180% on average.





Change in Resistivity from Untreated to Treated Surface



The resistivity at each location
within the treated surface
exceeded the low-to-high values of
the untreated surface
measurements in nearly all cases.



- The vast majority of the locations show an increase in the resistivity of the concrete deck.
- The inverse relationship between resistivity and corrosion rate provides excellent insight into the effectiveness of silane based SACI.







- Untreated areas <u>increased</u> by 0.58 µA/cm^{2.}
- The treated areas demonstrated virtually no change at all
- The corrosion inhibiting properties of silane based SACI had a significant impact on the corrosion rate in a relatively short period of time.

WHAT TO LOOK FOR...

- Proven Performance History
 - Lab
 - Field
- Compatibility
 - Overlays
 - Concrete Repair
- Ask: What does this do for me.....

Extension of Service Life Due to SACI Treatment



Based on concrete cover of 1" average and diffusion data from concrete

Estimated Retreat Times for SACI

